

3-[ $\alpha$ -(4-Methoxyphenyl)- $\beta$ -phenylvinyl]-2-methyl-1-phenylsulfonyl-1*H*-indoleG. Usha,<sup>a</sup> S. Selvanayagam,<sup>a</sup>  
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## Key indicators

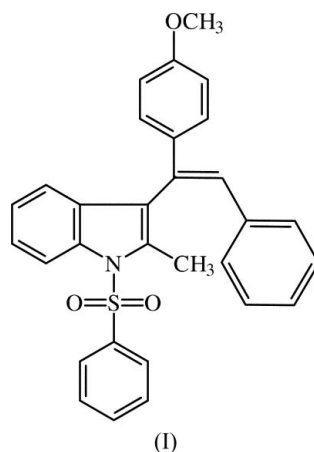
Single-crystal X-ray study  
 $T = 273$  K  
Mean  $\sigma(\text{C}-\text{C}) = 0.003$  Å  
 $R$  factor = 0.048  
 $wR$  factor = 0.142  
Data-to-parameter ratio = 18.1For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

In the title compound,  $\text{C}_{30}\text{H}_{25}\text{NO}_3\text{S}$ , the phenylsulfonyl and methoxyphenyl groups are almost perpendicular to the indole unit, whereas the other phenyl ring is inclined at an angle of  $68.9(1)^\circ$  to it. The benzene ring of the phenylsulfonyl substituent makes a dihedral angle of  $68.0(1)^\circ$  with the benzene ring of the methoxyphenyl substituent. The molecular packing in the crystal is stabilized by weak intra- and intermolecular  $\text{C}-\text{H}\cdots\text{O}$  interactions.

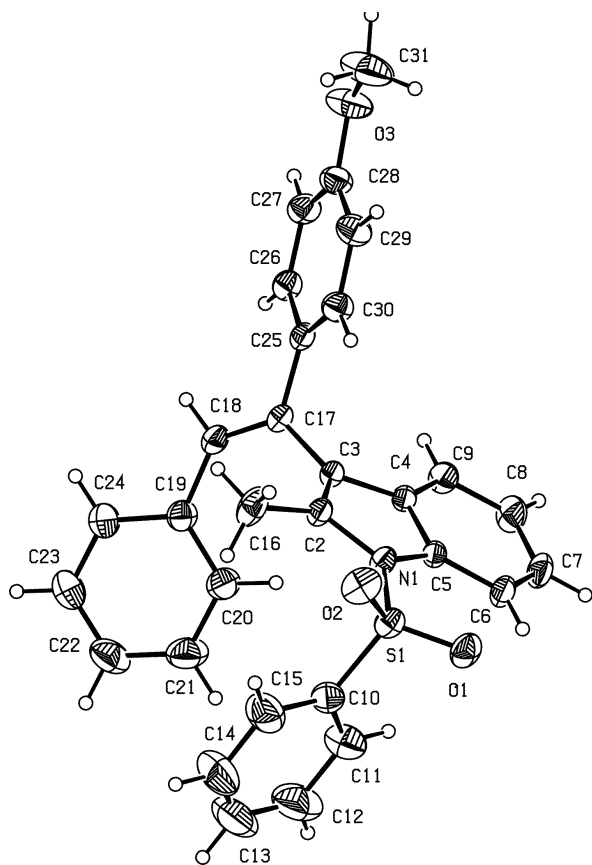
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## Comment

The indole unit is observed in plants (Nigović *et al.*, 2000). Indole-3-acetic acid is a naturally occurring plant growth hormone that controls a number of plant-growth activities (Fargasova, 1994). Many indole-containing natural products are found to exhibit psychotropic (Grinev *et al.*, 1978) and hypertensive (Merk, 1971) properties. A large number of biologically active compounds, mostly those affecting the central nervous system (Zhang & Liebeskind, 1996), contain indolines and their oxidized counterparts as important pharmacophores. Some of the indole derivatives possess antitumour (Schollmeyer *et al.*, 1995) and antibacterial (Okabe & Adachi, 1998) activities. The wide range of biological activities of indole and its derivatives prompted us to undertake this structural study of the title compound, (I).



The S—O, S—C and S—N bond distances are in good agreement with the reported values of 1.435 (5), 1.767 (7) and 1.685 (5) Å, respectively (Govindasamy *et al.*, 1998). The larger values of the C—N distances in the indole unit [C5—N1 = 1.409 (2) Å and C2—N1 = 1.422 (2) Å] are due to the electron-withdrawing character of the phenylsulfonyl group, and this phenomenon is observed in similar reported structures (Rodriguez *et al.*, 1995; Govindasamy *et al.*, 1997, 1998).



**Figure 1**  
The molecular structure of the title compound, showing 30% probability displacement ellipsoids.

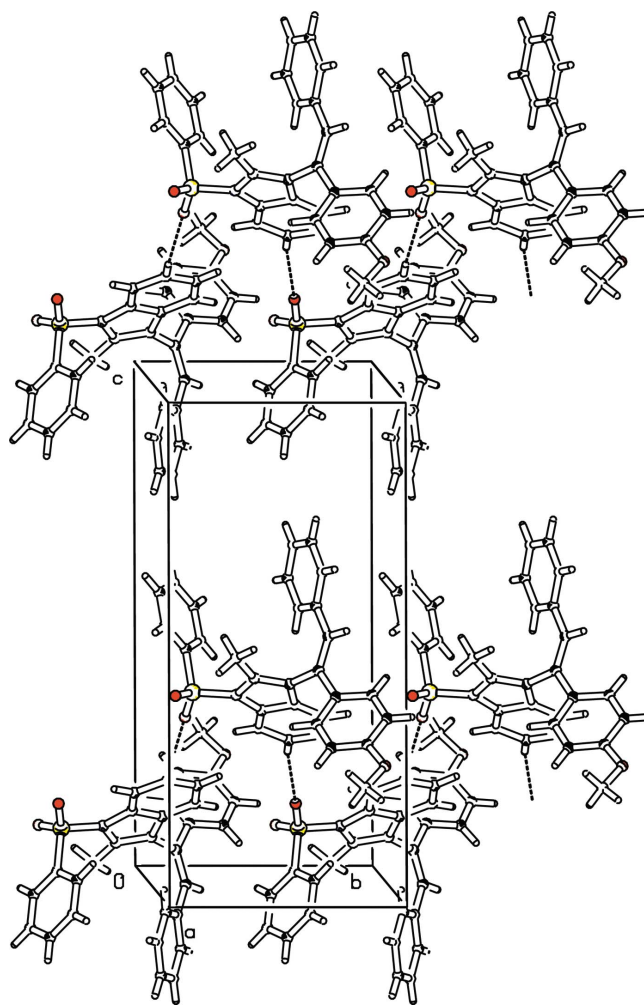
The sum of the angles around N1 ( $359.7^\circ$ ) shows the  $sp^2$ -hybridized character.

As normally observed in anisoles (Domiano *et al.*, 1979), the methoxy group atoms (O3 and C31) tend to be coplanar with the attached benzene ring but the exocyclic angles around C28 differ significantly [angle O3—C28—C29 is  $9.31(2)^\circ$  larger than O3—C28—C27]. The benzene rings of the phenylsulfonyl and methoxyphenyl substituents make angles of  $82.3(1)$  and  $82.7(1)^\circ$ , respectively, with the indole system. The angle between the other phenyl ring and the indole ring is  $68.9(1)^\circ$ .

There are weak intra- and intermolecular C—H $\cdots$ O interactions (Table 2 and Fig. 2).

## Experimental

To a stirred solution of sodium hydride (1.5 mmol) in dry tetrahydrofuran (THF, 5 ml) under an  $N_2$  atmosphere at reflux temperature (333 K) was added a solution of 3-[ $\alpha$ -(4-methoxyphenyl)- $\beta$ -phenylvinyl]-2-methyl-1*H*-indole (1 mmol) in dry THF (5 ml). After 1 h, the reaction mixture was cooled to room temperature. Phenylsulfonyl chloride (1.1 mmol) was added to the same solvent (5 ml) and stirred for 4 h. The solution was then poured over crushed ice and treated with a saturated solution of ammonium chloride, extracted with chloroform, concentrated, and passed through a chromatography column to give a pure white product. Diffraction-quality crystals were obtained from an ethyl acetate solution.



**Figure 2**  
The molecular packing of the title compound. Dashed lines indicate hydrogen bonds.

### Crystal data

$C_{30}H_{25}NO_3S$   
 $M_r = 479.57$   
Monoclinic,  $P2_1/c$   
 $a = 14.8973(11) \text{ \AA}$   
 $b = 8.9581(7) \text{ \AA}$   
 $c = 18.9941(14) \text{ \AA}$   
 $\beta = 97.658(1)^\circ$   
 $V = 2512.2(3) \text{ \AA}^3$   
 $Z = 4$

$D_x = 1.268 \text{ Mg m}^{-3}$   
Mo  $K\alpha$  radiation  
Cell parameters from 4749 reflections  
 $\theta = 2.2\text{--}27.2^\circ$   
 $\mu = 0.16 \text{ mm}^{-1}$   
 $T = 273(2) \text{ K}$   
Block, colourless  
 $0.22 \times 0.20 \times 0.18 \text{ mm}$

### Data collection

Bruker SMART CCD area-detector diffractometer  
 $\omega$  scans  
Absorption correction: none  
14866 measured reflections  
5748 independent reflections

4309 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.024$   
 $\theta_{\text{max}} = 28.0^\circ$   
 $h = -19 \rightarrow 19$   
 $k = -11 \rightarrow 11$   
 $l = -24 \rightarrow 20$

### Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.048$   
 $wR(F^2) = 0.143$   
 $S = 1.01$   
5748 reflections  
318 parameters  
H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0809P)^2 + 0.396P]$   
where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\text{max}} < 0.001$   
 $\Delta\rho_{\text{max}} = 0.28 \text{ e \AA}^{-3}$   
 $\Delta\rho_{\text{min}} = -0.33 \text{ e \AA}^{-3}$

**Table 1**  
Selected geometric parameters (Å, °).

S1—O2	1.416 (1)	O3—C28	1.368 (2)
S1—O1	1.417 (1)	O3—C31	1.395 (3)
S1—N1	1.666 (1)	N1—C5	1.409 (2)
S1—C10	1.750 (2)	N1—C2	1.422 (2)
O2—S1—O1	119.3 (1)	O2—S1—C10	109.2 (1)
O2—S1—N1	107.4 (1)	O1—S1—C10	109.0 (1)
O1—S1—N1	106.2 (1)	N1—S1—C10	104.8 (1)

**Table 2**  
Hydrogen-bond geometry (Å, °).

<i>D</i> —H··· <i>A</i>	<i>D</i> —H	H··· <i>A</i>	<i>D</i> ··· <i>A</i>	<i>D</i> —H··· <i>A</i>
C6—H6···O1	0.93	2.32	2.893 (2)	120
C7—H7···O1 <sup>i</sup>	0.93	2.53	3.397 (2)	155

Symmetry codes: (i)  $-x + 1, y + \frac{1}{2}, -z + \frac{1}{2}$ .

The H atoms were positioned geometrically and were treated as riding on their parent C atoms, with aromatic C—H = 0.93 Å, methyl C—H = 0.96 Å and methylene C—H = 0.97 Å, and with N—H = 0.86 Å, and with  $U_{\text{iso}} = 1.5U_{\text{eq}}(\text{C})$  for methyl H and  $1.2U_{\text{eq}}(\text{N,C})$  for the remaining H atoms.

Data collection: *SMART* (Bruker, 2001); cell refinement: *SAINTE* (Bruker, 2001); data reduction: *SAINTE*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP3* (Farrugia, 1997) and *PLATON* (Spek, 2003); software used to prepare material for publication: *SHELXL97* and *PARST* (Nardelli, 1995).

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